

**JEE CHEMISTRY**

*Topic: Radio Activity*

- Q.1** Radioactivity is a -
- (1) nuclear process                      (2) atomic process  
(3) chemical process                      (4) physical process
- Q.2** The value of decay constant of last element of radioactive series is -
- (1) infinite                      (2) much less  
(3) zero                      (4) equal to the decay constant of first element
- Q.3** If the pressure on a radioactive material is increased three times, then the mean life of the element -
- (1) does not change                      (2) will become three times  
(3) will become  $\frac{1}{3}$ rd                      (4) will depend on the initial pressure
- Q.4** A radioactive material emits 20  $\beta$ -particles per sec at 10°C. If the temperature is increased to 20°C then the emission rate of  $\beta$ -particles per sec is -
- (1) 20                      (2) 40  
(3) 30                      (4) 1
- Q.5** What will be the effect of dissolving a radioactive material in HNO<sub>3</sub>?
- (1) Its radioactive properties will remain unchanged  
(2) Its radioactive properties will change  
(3) The state of material cannot be predicted  
(4) None of these

- Q.6** The particles emitted by a radioactive substance are deflected in a magnetic field. The particle may be-
- (1) neutrons
  - (2) electrons
  - (3) protons
  - (4) hydrogen atoms
- Q.7** What will happen when a radioactive substance with mean life  $2 \times 10^5$  years is dissolved in  $H_2SO_2$  ?
- (1) it will dissociate into  $H^+$  and  $SO_2$  ions
  - (2) it will be converted into  $SO_2$  gas
  - (3) it will be converted into  $H_2$  gas
  - (4) it will remain unchanged
- Q.8** The half life of a radioactive material is 20 days. If it is heated to 10000 K, then its half life will become-
- (1)  $20 \times 10000$  days
  - (2)  $20/10000$  days
  - (3) 9800 days
  - (4) 20 days
- Q.9** The following is not an application of radioactive material -
- (1) to locate cracks in welding or castings
  - (2) to find the thickness of material
  - (3) in cigarette factory
  - (4) in photography
- Q.10** SI unit of radioactivity is -
- (1) curie
  - (2) rutherford
  - (3) rontgen
  - (4) bacqueral

**Q.11** The graph between remaining radioactive atoms and time for a radioactive decay is -

- (1) straight line      (2) parabola  
(3) exponential      (4) ellipse

**Q.12** Number of active atoms in m gram material is :

(M → atomic weight)

- (1)  $Mm \times 6.02 \times 10^{23}$   
(2)  $(M/m) \times 6.02 \times 10^{23}$   
(3)  $6.02 \times 10^{23}/Mm$   
(4)  $(m/M) \times 6.02 \times 10^{23}$

**Q.13** The activity of a radioactive element (decay constant  $\lambda$ ) becomes  $\frac{1}{3}$  of initial activity  $A_0$  in 9 years then the decay constant after 9 years will -

- (1)  $\lambda$                       (2)  $\lambda/3$   
(3)  $\lambda/9$                     (4)  $2\lambda/3$

**Q.14** A radioactive sample contains two elements P and Q. The mass of each is  $10^{-3}$ kg. The ratio of their atomic weights is 1 : 3. Their half lives are 4s and 8s respectively. The mass of P and Q after 16s will respectively be -

- (1)  $1.25 \times 10^{-5}$  kg and  $2.5 \times 10^{-4}$  kg  
(2)  $6.25 \times 10^{-5}$  kg and  $2.5 \times 10^{-4}$  kg  
(3)  $6.25 \times 10^{-5}$  kg and  $1.25 \times 10^{-4}$  kg  
(4)  $2.25 \times 10^{-5}$  kg and  $6.25 \times 10^{-4}$  kg

**Q.15** A fraction of  $\frac{5}{9}$  of a radioactive substance decays in time t. What fraction of the substance would had been active after time  $\frac{t}{2}$  -

- (1)  $1/2$                       (2)  $2/3$   
(3)  $3/4$                       (4)  $4/5$

- Q.16** What percentage of the atoms in a sample will remain undecayed in a time equal to mean life ?
- (1) 100%                      (2) 63%  
(3) 50 %                      (4) 37%
- Q.17** If the quantity of radioactive material reduces by 10% in 5 days, then the quantity that remains after 20 days will be -
- (1) 70%                      (2) 75 %  
(3) 65 %                      (4) 60%
- Q.18** The half life of a radioactive substance is 23.10 minute. If  $10^{23}$  atoms of the substance are active at any instant of time, then the activity of the substance will be - (in dps)
- (1)  $1 \times 10^{19}$                       (2)  $3 \times 10^{19}$   
(3)  $5 \times 10^{19}$                       (4)  $7 \times 10^{19}$
- Q.19** We get  $N_1$  and  $N_2$   $\beta$ -particles per second from two specimens of a radioactive specimen, then the ratio of number of atoms present in the samples is -
- (1)  $N_2/N_1$                       (2)  $N_1/N_2$   
(3)  $N_1^2/N_2^2$                       (4) None of these
- Q.20** A radio active substance has  $t_{1/2} = 60$  min. After 3 hrs, what percentage of radioactivity will remain -
- (1) 50%                      (2) 17.5%  
(3) 12.5%                      (4) 25%
- Q.21** When 64 gms of a radioactive element are carried from Jaipur to Jodhpur in 2 hours, then 1 gm of active element remains. The half life of the element is -
- (1) 2 hours                      (2) 30 minute  
(3) 20 minute                      (4) 1 hour



**Q.28** The counting rate observed from radioactivity source at  $t = 0$  second was 1600 counts per second and at  $t = 8$  seconds it was 100 counts per second. The counting rate observed, as counts per second at  $t = 6$  seconds will be -

- (1) 400                      (2) 300  
 (3) 200                      (4) 150

**Q.29** A radioactive sample at any instant has its disintegration rate 5000 disintegrations per minute. After 5 minutes, the rate is 1250 disintegrations per minute. Then, the decay constant (per minute) is -

- (1)  $0.8 \ln 2$               (2)  $0.4 \ln 2$   
 (3)  $0.2 \ln 2$               (4)  $0.1 \ln 2$

**Q.30** The fraction of a radioactive material which remains active after time  $t$  is  $9/16$ . The fraction which remains active after time  $t/2$  will be -

- (1)  $4/5$                       (2)  $7/8$   
 (3)  $3/5$                       (4)  $3/4$

## ANSWER KEY

<b>Que.</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>	<b>6</b>	<b>7</b>	<b>8</b>	<b>9</b>	<b>10</b>
<b>Ans.</b>	1	3	1	1	1	2	4	4	4	4
<b>Que.</b>	<b>11</b>	<b>12</b>	<b>13</b>	<b>14</b>	<b>15</b>	<b>16</b>	<b>17</b>	<b>18</b>	<b>19</b>	<b>20</b>
<b>Ans.</b>	3	4	1	2	2	4	3	3	2	3
<b>Que.</b>	<b>21</b>	<b>22</b>	<b>23</b>	<b>24</b>	<b>25</b>	<b>26</b>	<b>27</b>	<b>28</b>	<b>29</b>	<b>30</b>
<b>Ans.</b>	3	4	4	4	3	3	3	3	2	4