## JEE CHEMISTRY

## Topic : Aromatic compound

Q. 1 An organic compound contains about $7.7 \%$ carbon. Its acidic sodium extract gives a white precipitate with $\mathrm{AgNO}_{3}$. This precipitate is soluble in the excess of ammonia. The compound is
[1] $\mathrm{CHCl}_{3}$
[2] $\mathrm{CHBr}_{3}$
[3] $\mathrm{CHI}_{3}$
[4] $\mathrm{CCl}_{4}$
Q. 2 Match the following and pick up the correct answer

Compound
Use
$\begin{array}{lll}\mathrm{A} & \mathrm{CHCl}_{3} & \text { (i) Refrigerant }\end{array}$
B $\quad \mathrm{CCl}_{4}$
(ii) Fire extinguisher

C $\quad \mathrm{CF}_{2} \mathrm{Cl}_{2}$
(iii) Anaesthetic

The correct answer is
[1] A-(i), B-(ii), C-(iii)
[2] A-(iii), B-(ii), C-(i)
[3] A-(iii), B-(i), C-(ii)
[4] None of the above
Q. 3 Ethylidene bromide can be prepared by the following reaction
[1] Acetone with $\mathrm{PBr}_{3}$
[2] Acetone with $\mathrm{PBr}_{5}$
[3] Ethene with HBr
[4] Ethyne with HBr
Q. 4 A compound (A) is formed by the reaction of ethylene with bromine which on reacting with aqueous KOH gives a compound (B). The compound (B) can also be prepared by the reaction of ethylene with the following
[1] Baeyer's reagent
[2] Oxygen in the pesence of silver catalyst and then acidic hydrolysis
[3] performic acid and the product undergoes acidic hyrolysis
[4] All of the above
Q. 5 The product of the reaction of methyl magnesium bromide with methanol can also be prepared by the following reaction
[1] Reduction of methyl bromide
[2] Decarboxylation of sodium ethanoate
[3] Reduction of methyl alcohol
[4] All of the above
Q. 6 The product of the reaction of sodium acetylide with alkyl halide is
[1] A terminal alkyne
[2] A terminal alkene
[3] A non terminal alkyne
[4] [1] and [3] both
Q. 7 Carbon tetrachloride can be prepared by the following reaction
[1] $\mathrm{CS}_{2}$ with $\mathrm{Cl}_{2}$ in the presence of $\mathrm{I}_{2}$
[2] $\mathrm{CS}_{2}$ with $\mathrm{S}_{2} \mathrm{Cl}_{2}$ in the presence of $\mathrm{I}_{2}$
[3] $\mathrm{CHCl}_{3}$ with $\mathrm{Cl}_{2}$ in the presence of $\mathrm{I}_{2}$
[4] All of the above
Q. 8 The adduct of which of the following compounds with Grignard's reagent does not form a primary alcohol on hydrolysis
[1] $\mathrm{O}_{2}$
[2] Oxirane
[3] Methanal
[4] Ethanal
Q. 9 For the preparation of $\alpha$-butylene from methyl magnesium chloride, it will be reacted with
[1] Propene
[2] Propyl chloride
[3] 3-Chloro propene
[4] 2-Chloro propene
Q. 10 The vapour density of an organic compound is 23.0 . It contains $52.17 \% \mathrm{C}$ and $13 \% \mathrm{H}$. It gives iodoform test. The compound is
[1] Ethanol
[2] dimethyl ether
[3] Acetone
[4] Methanol
Q. 11 Identify Z in the following reaction
$\mathrm{CH}_{3} \mathrm{CN} \xrightarrow{\mathrm{Na} / \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}} \mathrm{X} \xrightarrow{\mathrm{HNO}_{2}} \mathrm{Y} \xrightarrow{\mathrm{KMnO}_{4}, \mathrm{H}^{+}} \mathrm{Z}$
[1] $\mathrm{CH}_{3} \mathrm{CHO}$
[2] $\mathrm{CH}_{3} \mathrm{CONH}_{2}$
[3] $\mathrm{CH}_{3} \mathrm{COOH}$
[4] $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{NHOH}$
Q. 12 Which of the following is a correct statement
[1] $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{Br}$ reacts with alcoholic KOH to give $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}$
[2] Reaction of $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{Br}$ with metallic Na gives ethane
[3] $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{Br}$ reacts with sodium ethoxide to give ether
[4] $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{Br}$ reacts with AgCN to give ethyl cyanide
Q. 13 The product of the reaction of chloromethane with sodium sulphide is
[1] Dimethyl sulphide
[2] Methan thiol
[3] Mixture of both the above
[4] None of the above
Q. 14 The compound with highest boiling point is -
[1] Ethylene bromide
[2] Ethylene chloride
[3] Ethylidene bromide
[4] Ethylidene chloride
Q. 15 The main product obtained by the electrolysis of the aqueous ethanolic solution of potassium bromide and sodium carbonate, is
[1] Ethyl bromide
[2] Bromoform
[3] Ethylene bromide
[4] Ethylidene bromide
Q. 16 Which of the following is Swart's reaction
$[1] \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{Cl}+\mathrm{AgF} \xrightarrow{\Delta} \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{~F}+\mathrm{AgCl}$
[2] $2 \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{Br}+2 \mathrm{Zn} \rightarrow\left(\mathrm{C}_{2} \mathrm{H}_{5}\right)_{2} \mathrm{Zn}+\mathrm{ZnBr}_{2}$
[3] $2 \mathrm{CHCl}_{3}+6 \mathrm{Ag} \rightarrow \mathrm{CH} \equiv \mathrm{CH}+6 \mathrm{AgCl}$
$[4] \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{Br}+\mathrm{NaI} \rightarrow \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{I}+\mathrm{NaBr}$
Q. 17 The chloroform exposed to air and sunlight gives white precipitate with $\mathrm{AgNO}_{3}$ solution because it contains
[1] Phosgene
[2] Hydrogen chloride
[3] Chlorine
[4] Mixture of all the above
Q. 18 The following type of compounds are obtained by the reaction of a carboxylic acid with lead tetra acetate and lithium chloride in benzene
[1] Alkyl halides
[2] Acid chlorides
[3] $\mathrm{CO}_{2}+\mathrm{H}_{2} \mathrm{O}$
[4] None of the above

## HALOBENZENE

Q. 19 Which of the following reactions is more suitable than the remaining three for obtaining iodobenzene ?
[1]

[2]

[3]

[4]

Q. 20 The reaction of chlorobenzene with which of the following reagents is not an example of electrophilic substitution reaction?
[1] $\mathrm{Cl}_{2}+$ Fe powder
[2] Conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$
[3] Conc. $\mathrm{HNO}_{3}+$ Conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$
[4] $\mathrm{CCl}_{3} \mathrm{CHO}+$ Conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$
Q. 21 At the time of preparation of chlorobenzene from benzne, which of the following can be used as a halogen carrier?
[1] A lewis acid
[2] Elements like iron, iodine etc
[3] A tertiary amine base
[4] All of the above
Q. 22 Which of the following can be obtained by Balz-Schiemann reaction?
[1] Fluorobenzene
[2] Chlororbenzene
[3] Bromobenzene
[4] lodobenzene
Q. 23 Which of the following reactions can be used to obtain chlorobenzene from benzenediazonium chloride ?
[A] Sandmeyer reaction
[2] Balz-Schemann reaction
[C] Rashing process
[4] Gattermann reaction
[1] A and B
[2] A, B and C
[3] A, C and D
[4] C and D
Q. 24 Which of the following catalysts is used in the preparation of chlorobenzene by Gattermann reaction ?
[1] $\mathrm{CuSO}_{4}$
[2] $\mathrm{CuCl}_{2}$
[3] $\mathrm{Cu}_{2} \mathrm{Cl}_{2}$
[4] Cu
Q. 25 Which of the following catalysts is used in the preparation of chlorobenzene by Sandmeyer's reaction?
[1] $\mathrm{CuCl}_{2}$
[2] $\mathrm{Cu}_{2} \mathrm{Cl}_{2}$
[3] $\mathrm{CuSO}_{4}$
[4] Cu
Q. 26 Which of the following compounds is obtained by Borodine-Hunsdiecker reaction of silver benzoate?
[1] Fluorobenzene
[2] Chlorobenzene
[3] Bromobenzene
[4] lodobenzene
Q. 27 Which of the following reagents is used for obtaining chlorobenzne from p-chlorophenol ?
[1] Zinc dust
[2] Soda lime
[3] Sodamide
[4] Copper powder
Q. 28 All of the following properties are exhibited by chlorobenzene, except :
[1] Almond-like faint smell
[2] Volatility
[3] Influammability
[4] Nonpoisonous nature
Q. 29 In Dow process, chlorobenzene is reacted with which of the following reagents?
$[\mathrm{A}] \mathrm{O}_{2}+\mathrm{HCl}$
[B] NaOH
[C] $\mathrm{H}_{2} \mathrm{O}$
[4] $\mathrm{Na}_{2} \mathrm{CO}_{3}$
[1] A and C
[2] A and D
[3] B and C
[4] B and D
Q. 30 Condensation of chlorobenzene and chloral hydrate is carried out in the presence of concentrated sulphuric acid for obtaining ?
[1] D.D.T.
[2] Chloropicrin
[3] B.H.C
[4] Dichlorodiphenylethane

## Answer $\mathbf{K}_{\text {ey }}$

| Qus. | $\mathbf{1}$ | $\mathbf{2}$ | $\mathbf{3}$ | $\mathbf{4}$ | $\mathbf{5}$ | $\mathbf{6}$ | $\mathbf{7}$ | $\mathbf{8}$ | $\mathbf{9}$ | $\mathbf{1 0}$ | $\mathbf{1 1}$ | $\mathbf{1 2}$ | $\mathbf{1 3}$ | $\mathbf{1 4}$ | $\mathbf{1 5}$ | $\mathbf{1 6}$ | $\mathbf{1 7}$ | 18 | 19 | $\mathbf{2 0}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Ans. | 4 | 2 | 4 | 4 | 4 | 1 | 4 | 4 | 3 | 1 | 3 | 3 | 1 | 1 | 2 | 1 | 2 | 1 | 1 | 4 |
| Qus. | 21 | 22 | 23 | 24 | 25 | 26 | 27 | 28 | 29 | 30 |  |  |  |  |  |  |  |  |  |  |
| Ans. | 4 | 1 | 3 | 4 | 2 | 3 | 1 | 4 | 4 | 1 |  |  |  |  |  |  |  |  |  |  |

